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1.17 understand the use of the term mole to represent the amount of substance 1.18 understand the term mole as the Avogadro number of particles (atoms, molecules, formulae, ions or electrons) in a substance 1.19 carry out mole calculations using relative atomic mass ( $A_r$ ) and relative formula mass ( $M_r$ )

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### **Objectives Vocabulary Key Equations**

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Example Exercise 9.1 Atomic Mass and Avogadro's Number The atomic mass of each element is listed below the symbol of the element in the periodic table: Cu = 63.55 amu, Hg = 200.59 amu, S = 32.07 amu, and He = 4.00 amu.

### **Example Exercise 9.1 Atomic Mass and Avogadro's Number**

Magnesium is a light metal used in the manufacture of aircraft, automobile wheels, and tools. How many moles of magnesium is  $1.25 \times 10$

### **10.1 The Mole: A Measure- ment of Matter - Useful Advice**

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A. Introduction to Chemistry, Atoms and Elements Importance of Chemistry Question: If cataclysmic event were to destroy all knowledge of science what would be the most important knowledge to pass on to future generations? Answer: Everything is made of Atoms. Atomic Theory is the central theme of chemistry and most important idea in science.

### **A. Introduction to Chemistry, Atoms and Elements**

170 Guided Reading and Study Workbook 4. Look at Figure 16.8 on page 481. Circle the letter of the best procedure for making a 0.50-molar (0.50M) solution in a 1.0-L volumetric flask. a. Add distilled water exactly to the 1.0-L mark, add 0.50 mol of solute, and

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Convert 3.00 moles  $\text{As}_2\text{S}_3$  to grams. 6. How many moles are represented by 11.5 g of  $\text{C}_2\text{H}_5\text{OH}$ ? 7. What is the mass of 9.30 moles of  $\text{SiH}_4$ ? 8. Convert  $8.00 \times 10^{20}$  molecules of  $\text{H}_2$  to moles. 9. How many atoms of tin are found in 3.50 moles of tin? 10. How many grams of tin are found in 3.50 moles of tin? Bonus: How many atoms of hydrogen are found ...

### **Worksheet: Mole Problems Name - Home | Georgia Public**

308 Chapter 11 The Mole CHAPTER 11 What You'll Learn You will use the mole and ... in chemistry. Relate a mole to common counting units. Convert moles to number of ... The answer is less than 10 moles, as predicted, and has the correct unit. Analyze the Problem

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